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It is known that olefins are formed by the action of boric acid on alcohols at 270-350°C [1, 2]. Proceeding from this, we proposed that N_1 -(1,4-butadien-2-yl)thymine would be formed under similar conditions from N_1 -(1,4-dihydroxy-2-butyl)thymine (I). In fact, we found that N_1 -(3-furanidyl)thymine (II) (65% yield) is sublimed when I is heated with boric acid at 300-325°. At higher temperatures (350-450°) we obtained a mixture of several products, from which a substance with a UV spectrum characteristic for divinyl derivatives (λ_{max} 215 nm and a shoulder at 260 nm, D_{215}/D_{260} =1.96) was isolated by extraction with pentane. Furanidylthymine II is stable under acid hydrolysis conditions, in contrast to N_1 -(2-furanidyl)-thymine, which decomposes on similar treatment.

EXPERIMENTAL

N₁-(3-Furanidyl)thymine (II). A solution of 14.05 mmole of I and 9.38 mmole of boric acid in 35 ml of dimethylformamide (DMFA) was refluxed for 4 h with slow removal of the DMFA by distillation. The residue was heated in vacuo (10 mm) at 170-180° for 45 min. The resulting solid white mass was heated in a vacuum sublimation apparatus (10 mm) while slowly raising the temperature to 300-325°. Sublimation product II was recrystallized from alcohol to give colorless crystals with mp 181-182° and R_f 0.54 [Silufol, chloroform-ethanol (9.2:0.8)] in 65% yield. UV spectrum: $\lambda_{\rm max}$ 271 nm, log ϵ 4.01. IR spectrum: ν 1050 (C-O), 1680 cm⁻¹ (C=O). PMR spectrum: δ 2.74 (singlet, CH=C), 4.95 (octet, > N-CH), 5.8-6.5 (multiplet, CH₂-O), 7.I-8.2 (multiplet, CH₂-C), and 8.21 ppm (broad singlet, CH₃).

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